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Key (example: Tc):

- Atomic number: 43
- Atomic mass (mean relative): (98.91)
- Electron configuration: [Kr] 4d^6 5s
- Oxydation states: 7
- Symbol: Tc
- Melting point [°C]: 2140
- Boiling point [°C]: 5030
- Density [g/cm^3], for gases [g/l] (0°C,1013mbar): 11.5
- Electronegativity: 1.9
- Radioactive: *

Period	Group	Atomic number	Symbol	Atomic mass	Electron configuration	Oxydation states	Melting point [°C]	Boiling point [°C]	Density	Electronegativity	Radioactive
1	1 (1A)	1	H	1.01	1s	1,-1	-259	-253	0.09	2.1	
1	18 (0)	2	He	4.00	1s^2		-	-269	018	-	
2	1 (1A)	3	Li	6.94	[He] 2s	1	181	1330	0.53	1.0	
2	2 (2A)	4	Be	9.01	[He] 2s^2	2	1277	2970	1.85	1.5	
2	13 (3B)	5	B	10.81	[He] 2s^2 2p	3	(2030)	2550	2.35	2.0	
2	14 (4B)	6	C	12.01	[He] 2s^2 2p^2	4,2,-4	(3550)	4830	2.2	2.5	
2	15 (5B)	7	N	14.01	[He] 2s^2 2p^3	5,4,3,2,-3	-210	-196	1.25	3.0	
2	16 (6B)	8	O	16.00	[He] 2s^2 2p^4	-2,-1	-219	-183	1.43	3.5	
2	17 (7B)	9	F	19.00	[He] 2s^2 2p^5	-1	-220	-188	1.7	4.0	
2	18 (0)	10	Ne	20.18	[He] 2s^2 2p^6		-249	-246	0.9	-	
3	1 (1A)	11	Na	22.99	[Ne] 3s	1	98	892	0.97	0.9	
3	2 (2A)	12	Mg	24.31	[Ne] 3s^2	2	650	1107	1.74	1.2	
3	13 (3B)	13	Al	26.98	[Ne] 3s^2 3p	3	660	2450	2.70	1.5	
3	14 (4B)	14	Si	28.09	[Ne] 3s^2 3p^2	4,-4	1410	2680	2.33	1.8	
3	15 (5B)	15	P	30.97	[Ne] 3s^2 3p^3	5,3,-3	44	280	1.82	2.1	
3	16 (6B)	16	S	32.06	[Ne] 3s^2 3p^4	6,4,2,-2	119	445	2.07	2.5	
3	17 (7B)	17	Cl	35.45	[Ne] 3s^2 3p^5	7,5,3,1,-1	-101	-35	3.2	3.0	
3	18 (0)	18	Ar	39.95	[Ne] 3s^2 3p^6		-189	-183	1.78	-	
4	1 (1A)	19	K	39.10	[Ar] 4s	1	64	760	0.86	0.8	
4	2 (2A)	20	Ca	40.08	[Ar] 4s^2	2	838	1440	1.55	1.0	
4	3 (3A)	21	Sc	44.96	[Ar] 3d 4s^2	3	1539	2730	3.0	1.3	
4	4 (4A)	22	Ti	47.88	[Ar] 3d^2 4s^2	4,3	1668	3260	4.54	1.5	
4	5 (5A)	23	V	50.94	[Ar] 3d^3 4s^2	5,4,3,2,0	1900	3450	6.1	1.5	
4	6 (6A)	24	Cr	52.00	[Ar] 3d^5 4s	6,3,2,0	1875	2200	7.19	1.6	
4	7 (7A)	25	Mn	54.94	[Ar] 3d^5 4s^2	7,6,4,3,2,0,-1	1245	2097	7.43	1.5	
4	8 (8)	26	Fe	55.85	[Ar] 3d^6 4s^2	6,3,2,0,-2	1536	3000	7.86	1.8	
4	9 (8)	27	Co	58.93	[Ar] 3d^7 4s^2	3,2,0,-1	1495	2900	8.9	1.9	
4	10 (8)	28	Ni	58.70	[Ar] 3d^8 4s^2	3,2,0	1453	2730	8.9	1.9	
4	11 (1B)	29	Cu	63.55	[Ar] 3d^{10} 4s	2,1	1083	2595	8.96	1.9	
4	12 (2B)	30	Zn	65.38	[Ar] 3d^{10} 4s^2	2	420	906	7.13	1.6	
4	13 (3B)	31	Ga	69.72	[Ar] 3d^{10} 4s^2 4p	3	30	2237	5.91	1.6	
4	14 (4B)	32	Ge	72.59	[Ar] 3d^{10} 4s^2 4p^2	4	937	2830	5.32	1.8	
4	15 (5B)	33	As	74.91	[Ar] 3d^{10} 4s^2 4p^3	5,3,-3	Subl.	-	5.72	2.0	
4	16 (6B)	34	Se	78.96	[Ar] 3d^{10} 4s^2 4p^4	6,4,-2	217	685	4.79	2.4	
4	17 (7B)	35	Br	79.90	[Ar] 3d^{10} 4s^2 4p^5	7,5,3,1,-1	-7	58	3.12	2.8	
4	18 (0)	36	Kr	83.80	[Ar] 3d^{10} 4s^2 4p^6		-157	-152	3.7	-	
5	1 (1A)	37	Rb	85.47	[Kr] 5s	1	39	688	1.53	0.8	
5	2 (2A)	38	Sr	87.62	[Kr] 5s^2	2	768	1380	2.6	1.0	
5	3 (3A)	39	Y	88.91	[Kr] 4d 5s^2	3	1509	2927	4.47	1.2	
5	4 (4A)	40	Zr	91.22	[Kr] 4d^2 5s^2	4	1852	3580	6.49	1.4	
5	5 (5A)	41	Nb	91.22	[Kr] 4d^4 5s	5,3	2468	4927	8.57	1.6	
5	6 (6A)	42	Mo	95.94	[Kr] 4d^5 5s	6,5,4,3,2,0	2610	5560	10.2	1.8	
5	7 (7A)	43	Tc	(98.91)	[Kr] 4d^6 5s	7	2140	5030	11.5	1.9	*
5	8 (8)	44	Ru	101.07	[Kr] 4d^7 5s	8,6,4,3,2,0,-2	2500	3900	12.4	2.2	
5	9 (8)	45	Rh	102.91	[Kr] 4d^8 5s	5,4,3,1,2,0	1966	3730	12.4	2.2	
5	10 (8)	46	Pd	106.42	[Kr] 4d^{10}	4,2,0	1552	3140	12.0	2.2	
5	11 (1B)	47	Ag	107.87	[Kr] 4d^{10} 5s	2,1	961	2210	10.5	1.9	
5	12 (2B)	48	Cd	112.41	[Kr] 4d^{10} 5s^2	2	321	765	8.65	1.7	
5	13 (3B)	49	In	114.82	[Kr] 4d^{10} 5s^2 5p	3	156	2080	7.31	1.7	
5	14 (4B)	50	Sn	118.69	[Kr] 4d^{10} 5s^2 5p^2	4,2	232	2270	7.30	1.8	
5	15 (5B)	51	Sb	121.75	[Kr] 4d^{10} 5s^2 5p^3	5,3,-3	631	1380	6.69	1.9	
5	16 (6B)	52	Te	127.60	[Kr] 4d^{10} 5s^2 5p^4	6,4,-2	450	990	6.24	2.1	
5	17 (7B)	53	I	126.90	[Kr] 4d^{10} 5s^2 5p^5	7,5,1,-1	114	183	4.94	2.5	
5	18 (0)	54	Xe	131.29	[Kr] 4d^{10} 5s^2 5p^6		-112	-108	5.89	-	
6	1 (1A)	55	Cs	132.91	[Xe] 6s	1	29	690	1.90	0.7	
6	2 (2A)	56	Ba	137.33	[Xe] 6s^2	2	714	1640	3.76	0.9	
6	3 (3A)	71	Lu	174.97	[Xe] 4f^{14} 5d 6s^2	3	1652	3327	9.84	1.2	
6	4 (4A)	72	Hf	178.49	[Xe] 4f^{14} 5d^2 6s^2	4	2222	5400	13.31	1.3	
6	5 (5A)	73	Ta	180.95	[Xe] 4f^{14} 5d^3 6s^2	5	2996	5425	16.5	1.5	
6	6 (6A)	74	W	183.85	[Xe] 4f^{14} 5d^4 6s^2	6,5,4,3,2,0	3410	5930	19.3	1.7	
6	7 (7A)	75	Re	186.21	[Xe] 4f^{14} 5d^5 6s^2	7,6,4,2,-1	3180	5900	21.0	1.9	
6	8 (8)	76	Os	190.20	[Xe] 4f^{14} 5d^6 6s^2	8,6,4,3,2,0,-2	3050	5500	22.6	2.2	
6	9 (8)	77	Ir	192.22	[Xe] 4f^{14} 5d^7 6s^2	6,4,3,2,1,0,-1	2454	4500	22.7	2.2	
6	10 (8)	78	Pt	195.08	[Xe] 4f^{14} 5d^9 6s	4,2,0	1769	3830	21.4	2.2	
6	11 (1B)	79	Au	196.97	[Xe] 4f^{14} 5d^{10} 6s	3,1	1063	2970	19.3	2.4	
6	12 (2B)	80	Hg	200.59	[Xe] 4f^{14} 5d^{10} 6s^2	2,1	-38	357	13.6	1.9	
6	13 (3B)	81	Tl	204.38	[Xe] 4f^{14} 5d^{10} 6s^2 6p	3,1	303	1457	11.85	1.8	
6	14 (4B)	82	Pb	207.20	[Xe] 4f^{14} 5d^{10} 6s^2 6p^2	4,2	327	1725	11.4	1.9	
6	15 (5B)	83	Bi	208.98	[Xe] 4f^{14} 5d^{10} 6s^2 6p^3	5,3	271	1560	9.8	1.9	
6	16 (6B)	84	Po	(209)	[Xe] 4f^{14} 5d^{10} 6s^2 6p^4	6,4,2	254	962	9.3	2.0	*
6	17 (7B)	85	At	(210)	[Xe] 4f^{14} 5d^{10} 6s^2 6p^5	7,5,3,1,-1	(302)	337	-	2.2	*
6	18 (0)	86	Rn	(222)	[Xe] 4f^{14} 5d^{10} 6s^2 6p^6		(-71)	-62	9.73	-	*
7	1 (1A)	87	Fr	(223.0)	[Rn] 7s	1	(27)	677	-	0.7	*
7	2 (2A)	88	Ra	226.03	[Rn] 7s^2	2	700	1140	5.0	0.9	*
7	3 (3A)	103	Lr	(262.1)	[Rn] 5f^{14} 6d 7s^2	3	-	-	-	-	*
7	4 (4A)	104	Rf	(261.1)	[Rn] 5f^{14} 6d^2 7s^2		-	-	-	-	*
7	5 (5A)	105	Db	(262.1)	[Rn] 5f^{14} 6d^3 7s^2		-	-	-	-	*
7	6 (6A)	106	Sg	(263.1)	[Rn] 5f^{14} 6d^4 7s^2		-	-	-	-	*
7	7 (7A)	107	Bh	(264.1)	[Rn] 5f^{14} 6d^5 7s^2		-	-	-	-	*
7	8 (8)	108	Hs	(265.1)	[Rn] 5f^{14} 6d^6 7s^2		-	-	-	-	*
7	9 (8)	109	Mt	(268)	[Rn] 5f^{14} 6d^7 7s^2		-	-	-	-	*
7	10 (8)	110	Uun	(269)	[Rn] 5f^{14} 6d^9 7s		-	-	-	-	*
7	11 (1B)	111	Uuu	(272)	[Rn] 5f^{14} 6d^{10} 7s		-	-	-	-	*
7	12 (2B)	112	Uub	(277)			-	-	-	-	*
7	14 (4B)	114	Uuq	(289)			-	-	-	-	*
7	16 (6B)	116	Uuh	(289)			-	-	-	-	*
7	18 (0)	118	Uuo	(293)			-	-	-	-	*
6	Lanthanoids	57	La	138.91	[Xe] 5d 6s^2	3	920	3470	6.17	1.1	
6	Lanthanoids	58	Ce	140.12	[Xe] 4f^2 6s^2	4,3	795	3468	6.67	1.1	
6	Lanthanoids	59	Pr	140.91	[Xe] 4f^3 6s^2	4,3	935	3127	6.77	1.1	
6	Lanthanoids	60	Nd	144.24	[Xe] 4f^4 6s^2	3	1024	3027	7.00	1.2	
6	Lanthanoids	61	Pm	(145)	[Xe] 4f^5 6s^2	3	(1027)	2460	7.22	-	*
6	Lanthanoids	62	Sm	150.36	[Xe] 4f^6 6s^2	3,2	1072	1790	7.54	1.2	
6	Lanthanoids	63	Eu	151.96	[Xe] 4f^7 6s^2	3,2	828	1439	5.26	-	
6	Lanthanoids	64	Gd	157.25	[Xe] 4f^7 5d 6s^2	3	1312	3000	7.89	1.1	
6	Lanthanoids	65	Tb	158.93	[Xe] 4f^9 6s^2	4,3	1356	2800	8.27	1.2	
6	Lanthanoids	66	Dy	162.50	[Xe] 4f^{10} 6s^2	3	1407	2600	8.54	-	
6	Lanthanoids	67	Ho	164.93	[Xe] 4f^{11} 6s^2	3	1461	2600	8.80	1.2	
6	Lanthanoids	68	Er	167.26	[Xe] 4f^{12} 6s^2	3	1497	2900	9.05	1.2	
6	Lanthanoids	69	Tm	168.93	[Xe] 4f^{13} 6s^2	3,2	1545	1727	9.33	1.2	
6	Lanthanoids	70	Yb	173.04	[Xe] 4f^{14} 6s^2	3,2	824	1196	6.98	1.1	
7	Actinoids	89	Ac	(227)	[Rn] 6d 7s^2	3	1050	-	10.1	1.1	*
7	Actinoids	90	Th	232.04	[Rn] 6d^2 7s^2	4	1750	3850	11.7	1.3	*
7	Actinoids	91	Pa	231.04	[Rn] 5f^2 6d 7s^2	5,4	(1230	-	15.4	1.4	*
7	Actinoids	92	U	238.03	[Rn] 5f^3 6d 7s^2	6,5,4,3	1132	3818	19.07	1.4	*
7	Actinoids	93	Np	(237.05)	[Rn] 5f^4 6d 7s^2	6,5,4,3	637	3900	19.5	1.3	*
7	Actinoids	94	Pu	(244)	[Rn] 5f^6 7s^2	6,5,4,3	640	3235	19.81	1.3	*
7	Actinoids	95	Am	(243)	[Rn] 5f^7 7s^2	6,5,4,3	994	-	13.7	1.3	*
7	Actinoids	96	Cm	(247)	[Rn] 5f^7 6d 7s^2	4,3	(1340)	3100	13.51	-	*
7	Actinoids	97	Bk	(247)	[Rn] 5f^9 7s^2	4,3	-	-	-	-	*
7	Actinoids	98	Cf	(251)	[Rn] 5f^{10} 7s^2	4,3	-	-	-	-	*
7	Actinoids	99	Es	(254)	[Rn] 5f^{11} 7s^2	3	-	-	-	-	*
7	Actinoids	100	Fm	(257)	[Rn] 5f^{12} 7s^2	3	-	-	-	-	*
7	Actinoids	101	Md	(258)	[Rn] 5f^{13} 7s^2	3	-	-	-	-	*
7	Actinoids	102	No	(259)	[Rn] 5f^{14} 7s^2	3,2	-	-	-	-	*

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Many mistakes on this file have been corrected in the meantime, thanks to error reports.

Thank you very much!

Updates

For updates to this printable version of the periodic table, visit http://periodictable.tsx.org. This file dates April 15th 2000.

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